Module 3

Chapter 5 – Microbial Metabolism

•	Metab	c and Anabolic Rea polism –				
•	Two g	eneral types of meta	bolic reactions	:		
	0		:			
	0		:			
•		I from Chapter 2:		whon	bond	_
	0	Energy can be		when	bond	5
	0	Energy can be		when	bond	S
•	Catab	olism –				
	0					
	0					
	0	_				
	0	Purpose is to		 		
		<u> </u>				
_	\ naha	 blism –		 		
•		DIISIII —				
	0					
	0					
		Requires	to	form bonds		
	0	Purpose is to				
•	Catab	olic reactions are "co	oupled with" an	abolic reactions		
	_	Coupled by				
	A met	abolic pathway is a _		of chemical		in a cell
•	Metab	olic reactions are co	ntrolled by pro	teins –		
_						
	zymes					
•	Enzyn	nes are biological cat	talysts			
	0					
_	0	for o		chamical reaction		
•		IUI a		_ chemical reaction		
•	Specii	ficity is due to	have characte	ristic		
	•					
•	If enzy	/me	enzyme does	 sn't	_	
				···· ·		
•		nes act on one or mo				
	•					

•	Each enzyme is specific
	0
	0
•	Names of enzymes end in ""
	 Sucrase, catalase, DNA polymerase
Fr	zyme components
•	Many enzymes are made entirely of
•	Some consist of 2 components
•	Apoprotein:
	Cofactor:
	•: organic cofactor
•	Holoenzyme:
_	
⊏r	zyme Mechanism
•	The general sequence of events in an enzymatic reaction
	Substrate(s) binds to
	This is called
	 The substrate(s) is transformed
	 Transformed molecule(s), the, released from enzyme
	No longer
	No longer Enzyme onlyto
•	Sequence continues until enzyme is, or enzyme
Fa	atora influencina anzymatia activity
•	Enzymes require to function
•	Hostile environments can cause proteins to
•	Temperature
	In general, chemical reactions speed up as temp
	For enzymes, too temperature causes
•	pH
	, pU io
	o All enzymes have pH
	o from preferred pH will
	Inhibitors
	Certain chemicals that bind to enzymes
	Octain chemicals that bind to cheymes
	Two classes of inhibitors:
•	
	 Competitive inhibitors – bind of the enzyment
	drugs inhibit (an essential
) synthesis via competitive inhibition
	drugs were the first effective antibacterial drugs
	Noncompetitive inhibitors – bind
	Akainhibition
	prevents replication of HIV via noncompetitive
	inhibition

Fe	edback inhibition
•	Control of enzymatic activity by use of
	0
En	ergy Production
•	Energy is stored in
	Recall, covalent bonds form by
•	Energy is stored in covalent bonds
•	Energy is stored in covalent bonds Catabolism involves stripping "high energy" from molecules and
	concentrate them in the bonds of ATP
•	Reactions that involve removing and adding electrons are called
	"" reactions
Ох	ridation-Reduction reactions
•	Oxidation:
•	Reduction:
•	reaction: an oxidation reaction with a reaction
•	reaction: an oxidation reaction with a reaction Catabolism is the of highly molecules
Th	e Generation of ATP
•	Energy released by redox reactions "trapped" by
•	Energy released by redox reactions "trapped" by of of
	o Addition of a
•	Requires energy3 types of to generate
Su	bstrate-level phosphorylation
•	of a PO4- to generates
•	process
•	and
Ox	idative phosphorylation
•	from organic molecule () used to
	generate () gradient
	 Gradient used to production
Ph	otophosphorylation
•	Occurs only in cells
•	transfers energy to electrons
	o get excited
•	Excited used to generate gradient, drive
	production
	 Similar to phosphorylation, using instead of
	to build dam
Me	etabolic pathways of energy production
•	Catabolism involves series of controlled reactions
	Releasing energy in one reaction generates too much heat
	Energy cannot be harnessed efficiently Outside Signature and a signature of the signa
•	Catabolism involves a series of reactions
	Catabolism involves a series of reactions o extracted to o Sequence of reactions called a
	O Sequence of reactions called a

		is performed by		
Carbohyd	rate catabolism			
 Carboh 	ydrates are	of energ	у	
0 _	is r	most common energy	source	
		two general processe		
	Cellular respiration	,		
	•			
o i	Fermentation			
	•			
	espiration			
•		(catabolism) of glucos	se phosphorylat	
0 1	Waste products are	, ,		
 Most of 	f the	is produced via	phosphorylat	ion
 Two type 	pes of respiration	· ·	,	
	•	piration – with		
0	Res	piration – without		
	e pathways involved			
Glycolysis	, , , , , , , , , , , , , , , , , , , ,			
	of alucose	molecule (6 Carbons) to 2 pyruvic acid molecules	(3 C)
	nd-products	(0 000000	, <u> </u>	()
	2			
0 2	- 2			
0 2	<u>-</u>	via		
Flectron	n Carriers		_	
		ctron carrier) removes	s electrons from	
_		-		
_	Becomes	. ((full elect	ron carrier)	
0	takes el	ectrons to their final d	estination	
The pathw				
	es 2 to get s	started		
•	is t	0		
• Overall	produce	d by alveolysis		
Glycoly		require oxygen		
Alternative	es to glycolysis			
	e-phosphate pathway	I		
			()
0 (Onerates	with alveolysis	_ (/
	Doudoroff pathway	with glycolyclo		
		and		
1 0	Does	and	•	
Intermedia	te Step	giyooiyala		
	•) is	and	
=	is produced			
	•			
•	waste			

Th	e Kreb	s Cycle				
•			or citric ac	id cycle		
•		es of 8 reaction steps	•			
•		oletely catabolizes organic mo	olecule to			
	0	,	_			
•	Main ı	products are .		1 ATP		
	0	products are, carriers		,		
•		end-product is same as starti				
Th	e path	•	Ü			
•		enters Krebs c	ycle			
•	Gene	rates				
•	1 ATF	produced by				
		generated as waste				
		tron Transport Chain				
•	A seri	es of				
•		pass el	ectrons to E	ETC		
	0	Become	_			
	0	Return to glycolysis, Krebs	cycle			
•	Energ	y released from	used to	o drive		_ from
	inside	cell to cell				
	0	Produced concer	ntration grad	dient –		
		Electrons end up on				
•	ETC g	generates	gradien	t		
•	g	radient favors	into d	cell		
	0	diff	use across	membrane t	freely	
•		re-enters cell via				
		Through transporter called			-	
•		captures ener				
0		Produces →				
5u		y of Respiration		- : ETO :		
•		oic respiration: the final electron		TINEICIS_		
	0	Organism is an		_		
	Λnaer	Oxygen is converted to robic respiration: the		_ /	in the ETC is	
	Allaci	I legally an				
	0	Usually an than cycle operates under	aerobic res	— niration beca	ause only part of	the Krehs
	O	cycle operates under	40.00.010	cond	litions	
•		sources can be u				
		Eg, can oxidize lipids,				
•		ners broken down by enzyme				
•	-	er subunits enter catabolism		points of alva	colvsis. Krebs cvc	ele
Fe	rmenta		- ,		J ,	-
•		tific definition:				
		Releases energy from		of	molecules	(food)
		Doesoxyg				•

	0	Does not use the				
	0	Does not use the Uses an		as the	 final electro	on acceptor
•	ATP (generated only via				·
•		converted into	organic n	 nolecule end	-product →	
	0	Main purpose is to rege	enerate	(f	rom) for
				•		,
	0	can partic	ipate in _		_ again	
•	Produ	ices	c	of energy		left in end-product
Tr	ie Path	•				
•		produce	es			
•		gl	lucose du	ring glycolysi	S	
•		produced d	uring glyc	olysis		
•		pyruvic	acid; rege	nerates		
•	Lactio	Acid fermentation produ	ıces		_	
•		nol fermentation produce:				
		herapy				
•		→ highly toxic				
		Inhibit Cytochrome c O	xidase, er	nzyme in		_
•		ETC				
	0	Cell cannot				
	0	Cell cannot				
	0	Cell not affe				
•		not affe	cted by cy	ranide		
Pł	notosy	ntnesis				
•	Photo	o: conversion of	energ	gy into	er	nergy ()
	0	depend	dent (ligh	t) reactions		
•	Synth					
		Carbon				molecules
•	Light	energy is absorbed by _				
	0	In	_ of chlord	plasts in euk	aryotes	
	0	In	_ of proka	ryotes		
•	Light	energizes or "		in		
•		ed electrons are passed	on to		_	
		ATP is generated				
•	Occui	rs in two ways:				
	0	()		
	0	()		
Cy	•	notophosphorylation				
•		elec				
•		passed on to _				
•	ATP (generated by				
•	Electr					
No	oncyclic	c photophosphorylation				
•		elec	trons			
•		passed on to				

•	ATP generated by
•	ATP generated by Electrons placed on → forms
•	Electrons in chlorophyll replaced by electrons from →
	produced as waste
_	ht-independent (Dark) reactions
0	Aka The cycle
0	Uses ATP, NADPH produced by noncyclic photophosphorylation to ""
0	from Process requires lots of
O Ma	produced etabolic diversity among organisms
IVIE	Organisms classified according to
•	Specifically, look at basic requirements to sustain life
	o
	0
•	Two classifications based on sources
	Phototrophs –
	Chemotrophs –
•	Two classifications based on sources
	 Autotrophs –
	■ Aka
	Heterotrophs –
	Aka
Cla	assification can be combined
•	Photoautotroph
	0
•	Photoheterotroph
	0
•	Chemoautotrophs
	0
•	Chemoheterotrophs
	0
Ch	emoheterotrophs
•	Heterotrophs classified according to their source of organic molecule
	 Saprophytes –
	o Parasites –
•	Metabolic diversity of great interest
	Can cause problems, provide potential solutions
	Rubber degrading bacteria can destroy gaskets in machinery RUT, can be used to degrade disparded tires.
	BUT, can be used to degrade discarded tires Bhodospecus anthropolis can cause disease in humans, animals.
	 Rhodococcus erythropolis can cause disease in humans, animals PLIT, can remove sulfur from netroloum
	 BUT, can remove sulfur from petroleum